Part 2 Properties of Matter

Name:

Part 2 Mass and Volume Lesson 1

Mass: The amount of ______ in an object. Weight has to do with gravity. On earth, mass and weight are the _____.

The standard unit of mass in the metric system is the _____.

Special Relationships

1 cubic meter of water has a mass of one ton, thus...

- 1 liter of water weighs 1 kilogram.
- 1 milliliter of water is one cubic centimeter.

Metric Ton: A cubic meter filled with _____ or 1,000 kilograms.

Area of Focus: Volume, Liter, I

Volume: The three-dimensional _____ an object occupies.

Volume is also the space that matter occupies.

Always measure a liquid at the bottom of the m_____

curve.



How to find the volume of a cube?



Find the volume of this cube?





Find the volume of this rectangle?



Find the volume of this rectangle? Each unit is 1cm³



How many milliliters is the toy scuba diver by using water displacement?



End Lesson 2 Mass and Volume

Part 2 Density Lesson 2

Density: How much ______ is contained in a given _____. We use grams/cm³ (grams per cubic centimeter) Density – mass d______ volume

> Mass D = ----- = grams/cm³ Volume

What is the density of this cube if it weighs 100 grams?



Answer = g/cm³

Please determine the densities of the following characters. Who is most dense?



Answer= Who is the most dense?

An object will float in Density of less than one =	
Density of more than one =	
What is the density of an objects whose mass	What is the density of an objects whose mass
is 500 grams and displaces 250 ml of water?	is 200 grams and displaces 250 ml of water?
Μ	Μ
Density = =	Density =
V	V
Will the object float in water? Yes / No	Will the object float in water? Yes / No

End Lesson 3 Density -Lesson 4 has no .doc component Part 2 Lesson 4 Density Continued

Density is defined as mass per unit \underline{V} . It is how much the mass is confined in a substance. It helps show if a ______ are packed closely together or spread far apart. To measure density, measure the mass on a balance, calculate volume and <u>d</u> the two. This process does not involve a chemical <u>c</u> and is thus a <u>p</u> property. Another way to measure density is by using its displacement of <u>l</u>.

Part 2 Lesson 5 Density Quiz, Lesson 7 is answers Volume and Density Quiz 1-10 From Slideshow / Video

Score out of 10____

1)	2)	3)
4)	5)	6)
7)	8)	9)
10)	*11) bonus	

What soda should we bring rafting and why?

6	Brand of Soda	Sodium mg	Calories	Sugar g	Mass g	Volume. ml	₽
						Can you find?	

Part 2 Properties of Matter

Part 2 Mass and Volume Lesson 1

Mass: The amount of <u>matter</u> in an object. Weight has to do with gravity. On earth, mass and weight are the <u>same.</u>

The standard unit of mass in the metric system is the gram.

Special Relationships

- 1 cubic meter of water has a mass of one ton, thus...
 - 1 liter of water weighs 1 kilogram.
 - 1 milliliter of water is one cubic centimeter.

Metric Ton: A cubic meter filled with <u>water</u> or 1,000 kilograms.

Area of Focus: Volume, Liter, I

Volume: The three-dimensional <u>space</u> an object occupies.

Volume is also the space that matter occupies.

 Always measure a liquid at the bottom of the <u>meniscus</u> curve.



How to find the volume of a cube?

Length x Width x Height = cm³



Find the volume of this cube?





Answer = 3 cm x 4 cm x 12 cm = 144 cm³

Find the volume of this rectangle? Each unit is 1cm³



How many milliliters is the toy scuba diver by using water displacement?

Answer= 5ml – 3.4ml Answer 1.6 ml



End Lesson 2 Mass and Volume Part 2 Density Lesson 2

Density: How much <u>space</u> is contained in a given <u>volume</u>. We use grams/cm³ (grams per cubic centimeter)

Density – mass <u>divided</u> volume

What is the density of this cube if it weighs 100 grams?



 $3^3 = 27 \text{ cm}^3 \text{ for volume}$

D = m/vMass = 100g 100g/27cm3 Answer = Density = 3.7 g/cm³

Please determine the densities of the following characters. Who is most dense?



Donkey Kong=	Yoshi=	Goomba=	Mario=
g/cm ³ . <mark>5g/cm3</mark>	g/cm³ <mark>1.625</mark>	g/cm³	g/cm³
	<mark>g/cm3</mark>	1.3 g/cm3	. <mark>8 g/cm3</mark>

Answer= Who is the most dense?

An object will float in water

- Density of less than one = will float in water
- Density of more than one = will sink in water

What is the density of an objects	What is the density of an objects
whose mass is 500 grams and	whose mass is 200 grams and
displaces 250 ml of water?	displaces 250 ml of water?
M 500g	M 200g
Density = = <mark>2g/cm³</mark>	Density = = . <mark>8 g/cm³</mark>
	V 250ml
V 250ml	
	Will the object float in water? Yes
Will the object float in water? No	/ No

End Lesson 3 Density -Lesson 4 has no .doc component

Part 2 Lesson 4 Density Continued

Density is defined as mass per unit <u>Volume</u>. It is how much the mass is confined in a substance. It helps show if <u>atoms</u> are packed closely together or spread far apart. To measure density, measure the mass on a balance, calculate volume and <u>divide</u> the two. This process does not involve a chemical <u>change</u> and is thus a <u>physical property</u>. Another way to measure density is by using its displacement of <u>liquid</u>.

Part 1B Lesson 6 Density Quiz, Lesson 7 is answers Volume and Density Quiz 1-10 From Slideshow / Video

Score out of 10_

1)	2)	3)
1 gram	1,000 gram or 1kg	53 ml
4)	5)	6)
24cm ³	3 Liters or 3,000 ml	4 ml
7)	8)	9)
mass D == grams/cm³ volume	Less than 1 g/cm ³	145 g / 125 cm³ = 1.16 g/cm³
10)	*11) bonus	
Honey, dishwashing more dense	Animal Jam	

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